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(54) **Conductive primers.**

(57) **An epoxyster linear block oligomer made with n-1 moles fatty acid units terminated with carboxylic acid groups and n moles of epoxy oligomer, when filled with carbon black, is a particularly effective electrically conductive primer for use over organic-based composites to be electrostatically painted.**

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TITLE

Conductive Primers

BACKGROUND

This invention concerns polymers and paints.

5 More particularly it concerns polyester-epoxy polymers filled with carbon black and particularly suited for use as electrically conductive coating compositions for use on automobiles, including non-metallic parts.

10 The present invention is an adaption of and improvement over that of U.S. application No.613,413 (FF-7795), Huybrechts, Meeus and Timmerman, filed May 24, 1984, the disclosure of which is incorporated herein by reference.

15 Coating systems for automobiles normally comprise a multiplicity of coatings applied to substrates of steel and sometimes also other materials, including electrically non-conductive organic-based composite materials. Typically, the steel is
20 treated with a phosphate, then a cathodic electrocoat primer is applied. A primer-surfacer is used next to smooth the surface and provide a thick enough coating to permit sanding to a smooth, flat finish. Then a topcoat system is applied, some-
25 times as a single colored coat, often as a basecoat with solid color or metallic pigments followed by a clear coat.

When a non-conductive material is used for part of a car body, it is important to provide
30 an electrically-conductive coating over it so that when the topcoat paint is applied electrostatically to the entire car, a uniform appearance will be

produced. Among such non-conductive materials are reaction injection molded (RIM) parts and various fiberglass filled composites.

It is difficult to find a paint that can
5 be made adequately conductive for such applications while still being tough and flexible enough to be a useful paint.

SUMMARY OF THE INVENTION

The present invention provides an electrically conductive coating composition comprising
10 (a) 5-40% by weight of a block-copolymerized linear oligomer having an average of 3 to 20 mole units of alternating units of

n moles of linear epoxy oligomers
15 terminated with oxirane groups on each end and an M_n in the range of 300-5000, and

(n-1) moles of fatty acid units terminated with carboxylic acid groups on each of two ends and having at least 9
20 carbon atoms,

said linear oligomer being terminated on each end with units of said epoxy oligomers,

(b) 1-20% of a blocked isocyanate which can be unblocked in the coating composition by heating
25 at temperatures not higher than 140°C,

(c) 1-30% of an aminoplast crosslinking agent,

(d) 0-40% by weight of pigment and extenders,

(e) 2-5% by volume based on binder solids of a conductive carbon black, and

30 (f) 20-80% by weight of organic solvents, having a specific conductivity measured as a dry, cured film of at least $5.10^{-3} (\Omega \text{ cm})^{-1}$.

Coated substrates are also part of the invention.

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DETAILED DESCRIPTION

With coating compositions of the invention, the carbon black content was varied and the conductivity of the resulting coating was measured

5 with a "Sprayability Meter" made by Ransburg Electro-Coating Corp. of Indianapolis, Indiana.

It was found that at least 2% carbon black by volume based on the total solids was needed to obtain electrostatic sprayability. The following
10 table shows the Ransburg Conductivity rating for various levels of carbon black. A conductivity of at least 100 is needed for a coated substrate to be amenable to electrostatic spraying.

| | <u>% Carbon Black Volume</u> | <u>Ransburg Conductivity</u> |
|----|----------------------------------|----------------------------------|
| 15 | 1 | 75 (not sprayable) |
| | 1.5 | 75 (not sprayable) |
| | 2 | 145 |
| | 2.5 | 165 |
| 20 | 3 | 165 |

Furthermore, the resins of the present invention exhibit substantial toughness, chip resistance and flexibility, making them suitable for use with carbon black fillings as conductive
25 coatings.

In the examples, parts and proportions are given by weight except where indicated otherwise.

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EXAMPLE 1Resin A - Block Epoxy Ester from
Epon 1001 and Pripol 1014 - dimer FA

| | | |
|----|---|--------------|
| | Epon 1001 epoxy resin (Shell) | 33.80 |
| 5 | Pripol 1014 dimerized fatty acid (Unilever) | 16.20 |
| | Triethylamine | 0.05 |
| | Methylisobutyl Ketone (MIBK) | 25 |
| | heat at reflux (114-116°C) until | |
| | acid number (AN) = 6-10 visc Q-T | |
| 10 | (Gardner-Holdt bubble viscosity) at | |
| | 50% solids in MIBK/CA blend = 1/1 | |
| | Cellosolve acetate (CA) | <u>24.95</u> |
| | | 100.00 |

The resulting resin was an oligomer containing 4
 15 moles of \overline{M}_n 900 epoxy and 3 moles of C_{36} dimerized
 fatty acid, for a total of 7 mole units, end
 terminated with epoxy.

EXAMPLE 2Conductive Primer for Coating Bulk Molding
Compound, Sheet Molding Compound and Polyamide

| | <u>Ingredient</u> | <u>Amount</u> | <u>Producer</u> |
|----|---------------------------|---------------|-----------------|
| | <u>Grinding stage</u> | | |
| | Block epoxy ester of Ex 1 | 29.73 | |
| | Solvesso 150 | 10.32 | Essochem |
| 25 | Bentone 34 | 0.33 | Nat'l Lead |
| | Carbon black XE-2 | | |
| | conductive carbon black | 1.11 | Phillips Petr. |
| | Xylene | 16.69 | |
| | Barytes (micronized) | 14.49 | Sachtleben |
| 30 | Zinc Oxide (American | 5.04 | De Craene |
| | process) | | |
| | Aluminum silicate pigment | 4.23 | |

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| | | | |
|---|---------------------------|-------------|------------------|
| | Titanium dioxide pigment | 11.79 | |
| | Guaiacol antioxidant | 0.67 | Rhone-Poulenc |
| | MPA 60X thixotropic agent | 0.67 | Baker Caster Oil |
| | Mineral spirits | <u>4.88</u> | Shell |
| 5 | | 100.00 | |

Let down stage

| | | | |
|----|---------------------------------------|-------------------|---------|
| | Grind | 86.44 | |
| | Maprenal MF590 melamine | 5.85 | Hoechst |
| | formaldehyde resin | | |
| 10 | Xylene | 1.00 | |
| | Blocked isocyanate | 3.84 | Bayer |
| | ketoxime-blocked aliphatic isocyanate | | |
| | Ethylene glycol butylether | 2.87 | |
| | acetate | <u> </u> | |
| 15 | | 100.00 | |

Reduce with Solvesso 150 to spray viscosity of 25-30 sec in Ford cup 4.

The resulting paint was sprayed onto non-conductive automobile parts and provided a tough coating which could be electrostatically sprayed with a topcoat.

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CLAIMS

1. An electrically conductive coating composition comprising

5 (a) 5-40% by weight of a block-copolymerized linear oligomer having an average of 3 to 20 mole units of alternating units of

n moles of linear epoxy oligomers terminated with oxirane groups on each end and an M_n in the range of 300-5000, and
10 (n-1) moles of fatty acid units terminated with carboxylic acid groups on each of two ends and having at least 9 carbon atoms,

said linear oligomer being terminated on each end
15 with units of said epoxy oligomers,

(b) 1-20% of a blocked isocyanate which can be unblocked in the coating composition by heating at temperatures not higher than 140°C,

(c) 1-30% of an aminoplast crosslinking agent,

20 (d) 0-40% by weight of pigment and extenders,

(e) 2-5% by volume based on binder solids of a conductive carbon black, and

(f) 20-80% by weight of organic solvents, having a specific conductivity measured as a dry,
25 cured film of at least $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.

2. The coating composition of claim 1 wherein the block epoxy ester is the reaction product of a bisphenol A diepoxy resin with a C_{36} -based aliphatic diacid.

30 3. The coating composition of claim 1 wherein the blocked isocyanate is a ketoxime-blocked aliphatic isocyanate.

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4. The coating composition of claim 1 wherein the crosslinking agent is selected from the group consisting of melamine formaldehyde resins, urea formaldehyde resins and benzoguanamine formaldehyde resins.

5. The coating composition of claim 4 wherein the crosslinking agent is a reactive partially alkylated melamine formaldehyde resin.

6. A coated substrate comprising a substrate of an organic-based composite material coated with a cured coating composition of claim 1, giving an electrical conductivity of at least $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.

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European Patent
Office

EUROPEAN SEARCH REPORT

0179281
Application number

EP 85 11 1936

| DOCUMENTS CONSIDERED TO BE RELEVANT | | | |
|---|---|--|---|
| Category | Citation of document with indication, where appropriate, of relevant passages | Relevant to claim | CLASSIFICATION OF THE APPLICATION (Int. Cl. 4) |
| A | EP-A-0 024 811 (MOBIL OIL) | | C 09 D 5/24 C 08 G 18/42 C 08 G 18/58 C 08 G 18/40 |
| A | US-A-4 380 596 (G.R. WATCHKO) | | C 09 D 3/58 C 09 D 3/49 C 08 K 3/04 |
| | | | TECHNICAL FIELDS SEARCHED (Int. Cl. 4) |
| | | | C 09 D C 08 G C 08 K |
| The present search report has been drawn up for all claims | | | |
| Place of search THE HAGUE | | Date of completion of the search 21-01-1986 | Examiner VAN PUymbROECK M.A. |
| CATEGORY OF CITED DOCUMENTS | | | |
| X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document | | T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document | |

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EUROPEAN PATENT SPECIFICATION

④⑤ Date of publication of patent specification: 11.01.89

②① Application number: 85111936.2

②② Date of filing: 20.09.85

⑥① Int. Cl.⁴: **C 09 D 5/24, C 08 G 18/42,
C 08 G 18/58, C 08 G 18/40,
C 09 D 3/58, C 09 D 3/49,
C 08 K 3/04**

⑤④ **Conductive primers.**

③① Priority: 21.09.84 US 652859

④③ Date of publication of application:
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④⑤ Publication of the grant of the patent:
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⑥④ Designated Contracting States:
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⑥⑤ References cited:
EP-A-0 024 811
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Courier Press, Leamington Spa, England.

Description

Background

This invention concerns polymers and paints. More particularly it concerns polyester-epoxy polymers filled with carbon black and particularly suited for use as electrically conductive coating compositions for use on automobiles, including non-metallic parts.

The present invention is an adaptation of and improvement over that of U.S. Patent 4,602,053, the disclosure of which is incorporated herein by reference.

Coating systems for automobiles normally comprise a multiplicity of coatings applied to substrates of steel and sometimes also other materials, including electrically non-conductive organic-based composite materials. Typically, the steel is treated with a phosphate, then a cathodic electro-coat primer is applied. A primer-surface is used next to smooth the surface and provide a thick enough coating to permit sanding to a smooth, flat finish. Then a topcoat system is applied, sometimes as a single colored coat, often as a base-coat with solid color or metallic pigments followed by a clear coat.

When a non-conductive material is used for part of a car body, it is important to provide an electrically-conductive coating over it so that when the topcoat paint is applied electrostatically to the entire car, a uniform appearance will be produced. Among such non-conductive materials are reaction injection molded (RIM) parts and various fiberglass filled composites.

It is difficult to find a paint that can be made adequately conductive for such applications while still being tough and flexible enough to be a useful paint.

Summary of the Invention

The present invention provides an electrically conductive coating composition comprising

(a) 5—40% by weight of a block-copolymerized linear oligomer having an average of 3 to 20 mole units of alternating units of

n moles of linear epoxy oligomers terminated with oxirane groups on each end and an \overline{M}_n in the range of 300—5000, and

(n-1) moles of fatty acid units terminated with carboxylic acid groups on each of two ends and having at least 9 carbon atoms,

said linear oligomer being terminated on each end with units of said epoxy oligomers,

(b) 1—20% of a blocked isocyanate which can be unblocked in the coating composition by heating at temperatures not higher than 140°C,

(c) 1—30% of an aminoplast crosslinking agent,

(d) 0—40% by weight of pigment and extenders,

(e) 2—5% by volume based on binder solids of a conductive carbon black, and

(f) 20—80% by weight of organic solvents, having a specific conductivity measured as a dry, cured film of at least $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.

Coated substrates are also part of the invention.

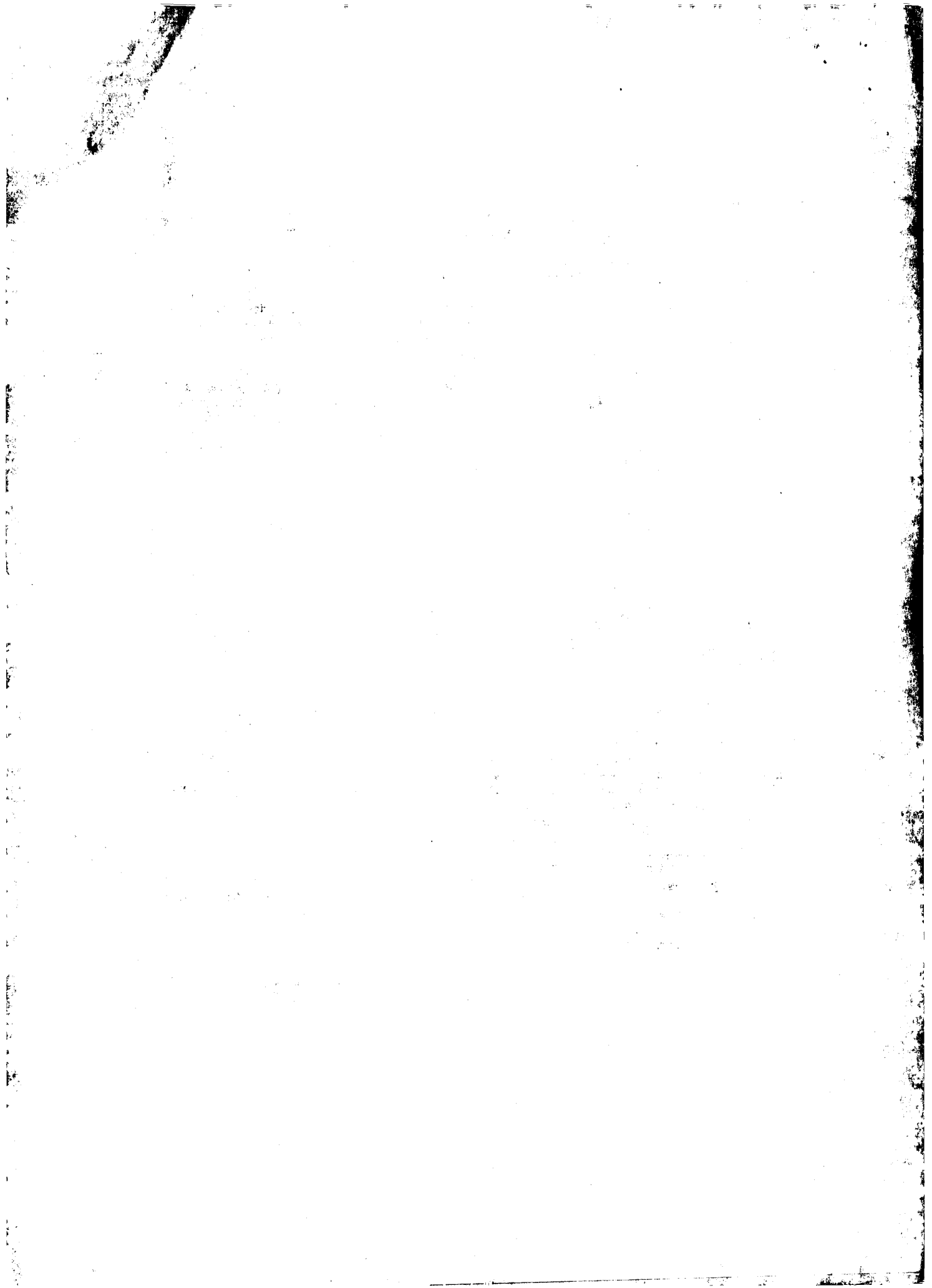
A preferred block epoxy ester is the reaction product of a bisphenol A diepoxy resin with a C_{36} -based aliphatic diacid and a preferred blocked isocyanate is a ketoxime-blocked aliphatic isocyanate.

The crosslinking agent is preferably selected from the group consisting of melamine formaldehyde resins, urea formaldehyde resins and benzoguanamine formaldehyde resins. The most preferred crosslinking agent is a reactive partially alkylated melamine formaldehyde resin.

Detailed Description

With coating compositions of the invention, the carbon black content was varied and the conductivity of the resulting coating was measured with a "Sprayability Meter" made by Ransburg Electro-Coating Corp. of Indianapolis, Indiana. It was found that at least 2% carbon black by volume based on the total solids was needed to obtain electrostatic sprayability. The following table shows the Ransburg Conductivity rating for various levels of carbon black. A conductivity of at least 100 is needed for a coated substrate to be amenable to electrostatic spraying.

| | % Carbon Black Volume | Ransburg Conductivity |
|----|--------------------------|-----------------------|
| 55 | 1 | 75 (not sprayable) |
| | 1.5 | 75 (not sprayable) |
| 60 | 2 | 145 |
| | 2.5 | 165 |
| | 3 | 165 |
| 65 | 2 | 145 |



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Furthermore, the resins of the present invention exhibit substantial toughness, chip resistance and flexibility, making them suitable for use with carbon black fillings as conductive coatings. In the examples, parts and proportions are given by weight except where indicated otherwise.

Example 1

Resin A — Block Epoxy Ester from
Epon® 1001 and Pripol® 1014 — dimer FA

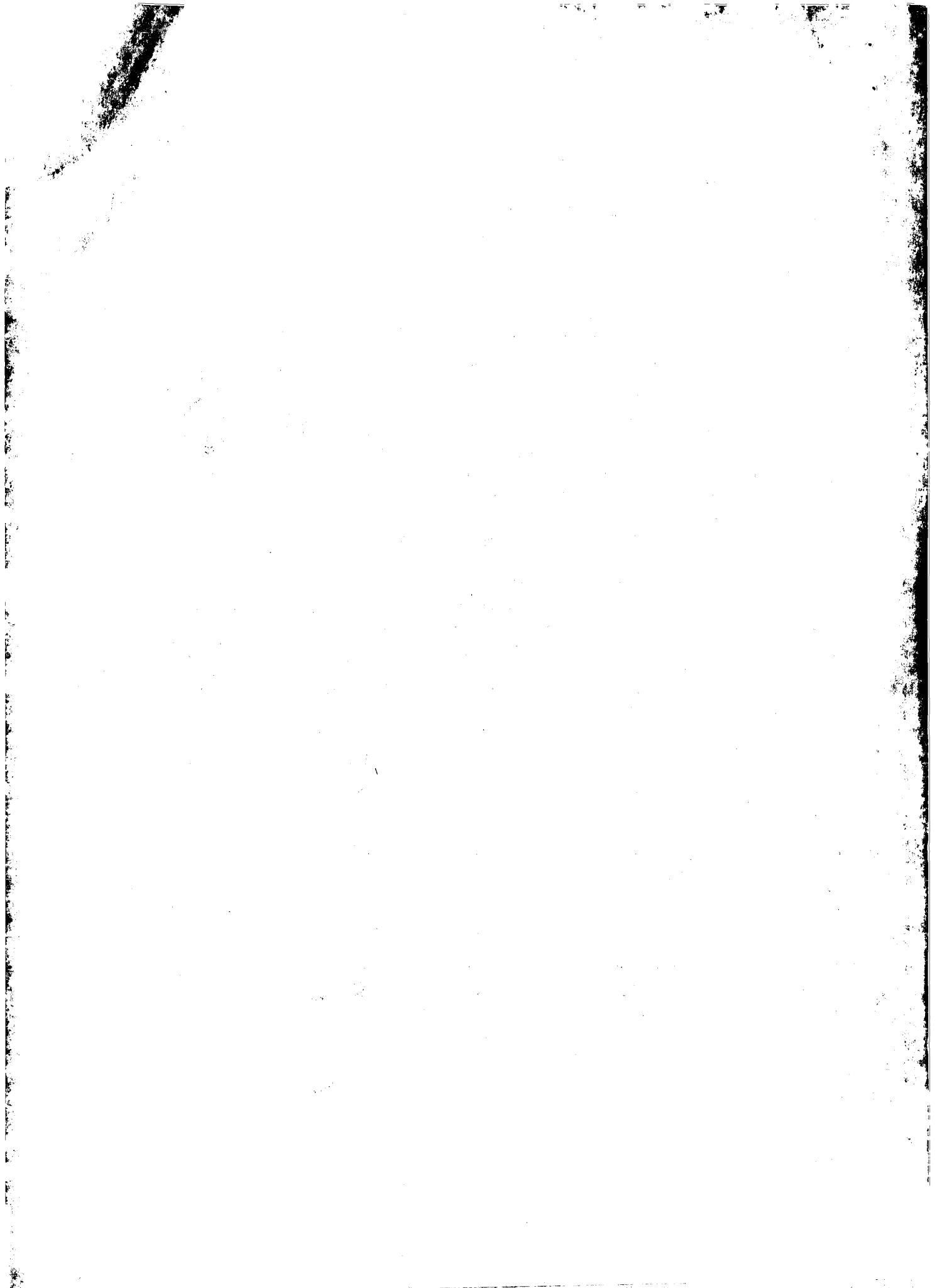
| | | |
|----|--|--------|
| 5 | | |
| 10 | Epon® 1001 epoxy resin (Shell) | 33.80 |
| | Pripol® 1014 dimerized fatty acid (Unilever) | 16.20 |
| 15 | Triethylamine | 0.05 |
| | Methylisobutyl Ketone (MIBK) | 25 |
| 20 | heat at reflux (114—116°C) until acid number (AN)=6—10 visc Q—T (Gardner-Holdt bubble viscosity) at 50% solids in MIBK/CA blend=1/1 | |
| 25 | Cellosolve® acetate (CA) | 24.95 |
| | | 100.00 |

The resulting resin was an oligomer containing 4 moles of Mn 900 epoxy and 3 moles of C₃₆ dimerized fatty acid, for a total of 7 mole units, end terminated with epoxy.

Example 2

Conductive Primer for Coating Bulk Molding
Compound, Sheet Molding Compound and Polyamide

| | | | |
|----|--|--------|------------------|
| 35 | Ingredient | Amount | Producer |
| | <u>Grinding stage</u> | | |
| 40 | Block epoxy ester of Example 1 | 29.73 | |
| | Solvesso® 150 | 10.32 | Essochem |
| 45 | Bentone® 34 | 0.33 | Nat'l Lead |
| | Carbon black XE—2 conductive carbon black | 1.11 | Phillips Petr. |
| 50 | Xylene | 16.69 | |
| | Barytes (micronized) | 14.49 | Sachtleben |
| | Zinc Oxide (American process) | 5.04 | De Craene |
| 55 | Aluminum silicate pigment | 4.23 | |
| | Titanium dioxide pigment | 11.79 | |
| 60 | Guaiacol® antioxidant | 0.67 | Rhone-Poulenc |
| | MPA 60X thixotropic agent | 0.67 | Baker Caster Oil |
| 65 | Mineral spirits | 4.88 | Shell |
| | | 100.00 | |



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| | | |
|----|--|--------------------|
| | <u>L t-down stage</u> | |
| | Grind | 86.44 |
| 5 | Maprenal® MF590 melamine formaldehyde resin | 5.85 Hoechst |
| | Xylene | 1.00 |
| 10 | Blocked isocyanate ketoxime-blocked aliphatic isocyanate | 3.84 Bayer |
| | Ethylene glycol butylether acetate | 2.87 |
| | | <hr/> 100.00 <hr/> |
| 15 | | |

Reduce with Solvesso® 150 to spray viscosity of 25—30 sec in Ford cup 4.

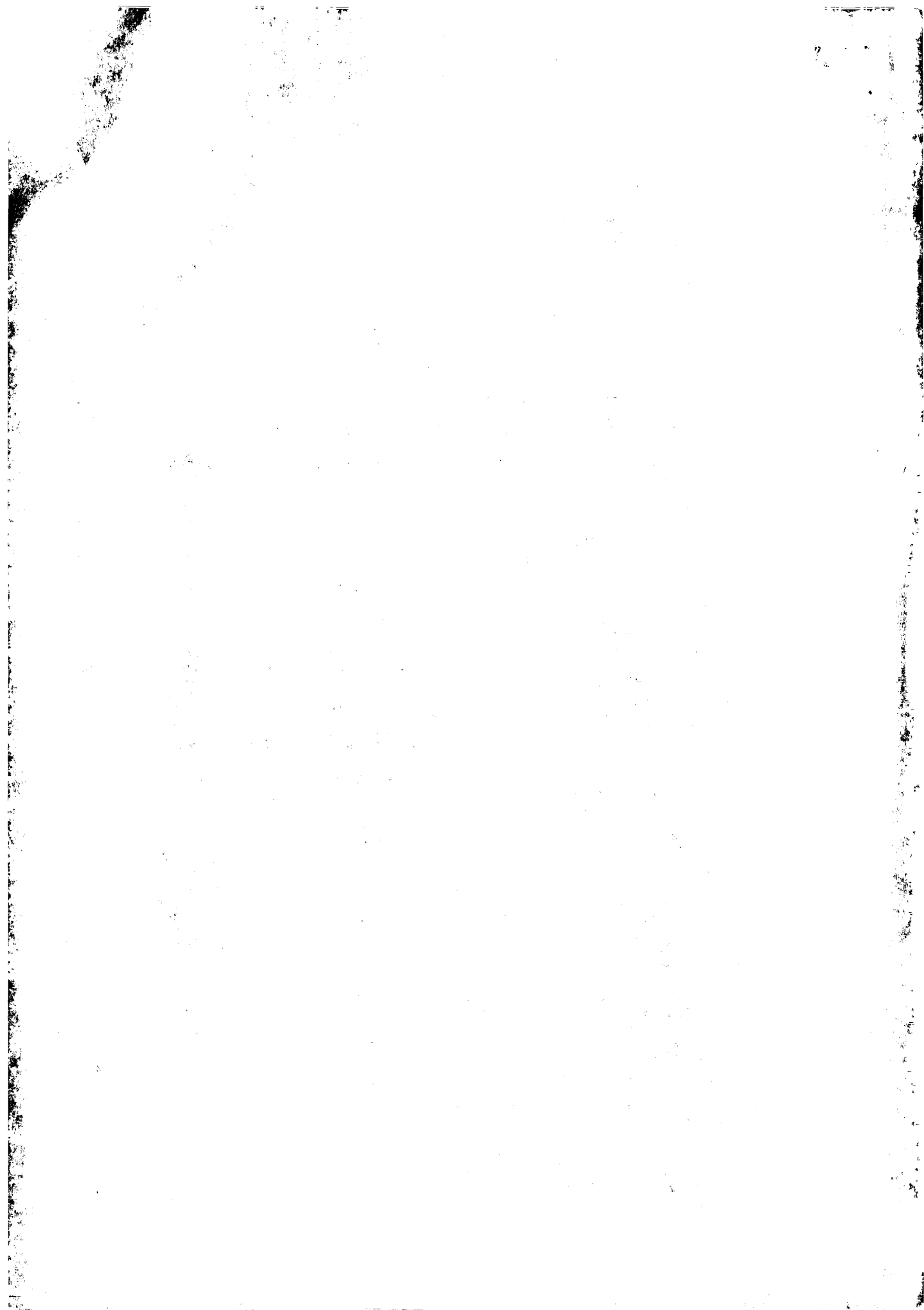
The resulting paint was sprayed onto non-conductive automobile parts and provided a tough coating which could be electrostatically sprayed with a topcoat.

Claims

1. An electrically conductive coating composition comprising
 - (a) 5—40% by weight of a block-copolymerized linear oligomer having an average of 3 to 20 mole units of alternating units of
 - n moles of linear epoxy oligomers terminated with oxirane groups on each end and an M_n in the range of 300—5000, and
 - (n-1) moles of fatty acid units terminated with carboxylic acid groups on each of two ends and having at least 9 carbon atoms,
 said linear oligomer being terminated on each end with units of said epoxy oligomers,
 - (b) 1—20% of a blocked isocyanate which can be unblocked in the coating composition by heating at temperatures not higher than 140°C,
 - (c) 1—30% of an aminoplast crosslinking agent,
 - (d) 0—40% by weight of pigment and extenders,
 - (e) 2—5% by volume based on binder solids of a conductive carbon black, and
 - (f) 20—80% by weight of organic solvents, having a specific conductivity measured as a dry, cured film of at least $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.
2. The coating composition of claim 1 wherein the block epoxy ester is the reaction product of a bisphenol A diepoxy resin with a C_{36} -based aliphatic diacid.
3. The coating composition of claim 1 wherein the blocked isocyanate is a ketoxime-blocked aliphatic isocyanate.
4. The coating composition of claim 1 wherein the crosslinking agent is selected from the group consisting of melamine formaldehyde resins, urea formaldehyde resins and benzoguanamine formaldehyde resins.
5. The coating composition of claim 4 wherein the crosslinking agent is a reactive partially alkylated melamine formaldehyde resin.
6. A coated substrate comprising a substrate of an organic-based composite material coated with a cured coating composition of claim 1, giving an electrical conductivity of at least $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.

Patentansprüche

1. Elektrisch leitende Überzugszusammensetzung, welche umfaßt
 - (a) 5—40 Gew.% eines blockcopolymerisierten linearen Oligomeren mit durchschnittlich 3 bis 20 Mol Einheiten an alternierenden Einheiten von
 - n-Molen von linearen Epoxyoligomeren, die an jedem Ende mit Oxirangruppen abgeschlossen sind und ein M_n im Bereich von 300—5000 aufweisen, und
 - (n-1)-Molen an Fettsäureeinheiten, die an jedem der beiden Enden mit Carbonsäuregruppen abgeschlossen sind und wenigstens 9 Kohlenstoffatome besitzen,
 w bei die linearen Oligomeren an jedem Ende mit Einheiten der genannten Epoxyoligomeren abgeschlossen sind,
 - (b) 1—20% eines blockierten Isocyanats, das in der Überzugszusammensetzung in den unblockierten Zustand überführt werden kann durch Erhitzen auf Temperaturen nicht höher als 140°C,
 - (c) 1—30% eines Aminoplastvernetzungsmittels,
 - (d) 0—40 Gew.-% Pigment und Streckmittel,



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- (e) 2—5 Volumen-%, bezogen auf den Gehalt an Binderfeststoff, eines leitfähigen Rußes, und
(f) 20—80 Gew.-% organische Lösungsmittel, und eine spezifische Leitfähigkeit, gemessen an einer trockenen, gehärteten Folie, von wenigstens $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$ aufweist.
2. Überzugszusammensetzung nach Anspruch 1, bei welcher der Blockepoxyester ein Reaktionsprodukt eines Bisphenol-A-diepoxyharzes mit einer aliphatischen Disäure auf C_{38} -Basis ist.
3. Überzugszusammensetzung nach Anspruch 1, bei der das blockierte Isocyanat ein ketoxim-blockiertes aliphatisches Isocyanat ist.
4. Überzugszusammensetzung nach Anspruch 1, bei der das Vernetzungsmittel ausgewählt ist aus der Gruppe bestehend aus Melaminformaldehydharzen, Harnstoffformaldehydharzen und Benzoguanaminformaldehydharzen.
5. Überzugszusammensetzung nach Anspruch 4, bei der das Vernetzungsmittel ein reaktives, teilweise alkyliertes Melaminformaldehydharz ist.
6. Überzogenes Substrat, umfassend ein Substrat aus einem Verbundmaterial auf organischer Basis, das mit einer gehärteten Zusammensetzung nach Anspruch 1 überzogen ist und eine elektrische Leitfähigkeit von wenigstens $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$ ergibt.

Revendications

1. Une composition de revêtement électriquement conducteur, comprenant
- (a) 5 à 40% en poids d'un oligomère linéaire copolymérisé en séquences comptant en moyenne 3 à 20 unités molaires de motifs alternés de
- n moles d'oligomères époxydes linéaires terminés à chaque extrémité par des groupes oxiranne et dont le M_n est compris dans l'intervalle de 300 à 5000, et
- $(n-1)$ moles de motifs acide gras terminés à chacune des deux extrémités par des groupes acide carboxylique et comptant au moins 9 atomes de carbone,
- ledit oligomère linéaire étant terminé à chaque extrémité par des motifs desdits oligomères époxydes,
- (b) 1 à 20% d'un isocyanate bloqué qui peut être débloqué dans la composition de revêtement par chauffage à des températures non supérieures à 140°C ,
- (c) 1 à 30% d'un agent réticulateur aminoplaste,
- (d) 0 à 40% en poids de pigment et de charges inertes,
- (e) 2 à 5% en volume, sur la base des solides liants, d'un noir de carbone conducteur, et
- (f) 20 à 80% en poids de solvants organiques, ayant une conductivité spécifique, mesurée à l'état de pellicule durcie sèche, d'au moins $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.
2. La composition de revêtement de la revendication 1, dans laquelle l'ester époxyde séquencé est le produit de réaction d'une résine diépoxyde de bisphénol A avec un diacide aliphatique en C_{38} .
3. La composition de revêtement de la revendication 1, dans laquelle l'isocyanate bloqué est un isocyanate aliphatique bloqué par une cétoxime.
4. La composition de revêtement de la revendication 1, dans laquelle l'agent réticulateur est choisi dans le groupe formé par les résines mélamine-formaldéhyde, les résines urée-formaldéhyde et les résines benzoguanamineformaldéhyde.
5. La composition de revêtement de la revendication 4, dans laquelle l'agent réticulateur est une résine mélamine-formaldéhyde réactive partiellement alkylée.
6. Un substrat revêtu comprenant un substrat en un matériau composite à base organique revêtu par une composition de revêtement durcie de la revendication 1, présentant une conductivité électrique d'au moins $5 \cdot 10^{-3} (\Omega \text{ cm})^{-1}$.

